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Solution This problem gives us, for a hypothetical ion pair, values for r_0 (0.35 nm), E_0 (6.13 eV), and n (10), and asks that we determine explicit expressions for attractive and repulsive energies of Equations 2.8 and 2.9. In essence, it is necessary to compute the values of A and B in these equations.

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Solution The average atomic weight of silicon (A_{Si}) is computed by adding fraction-of-occurrence/atomic weight Si products for the three isotopes—i.e., using Equation 2.2. (Remember: fraction of occurrence is equal to the percent of occurrence divided by 100.) Thus, $A_{Si} = f_{28} A_{28} + f_{29} A_{29} + f_{30} A_{30}$ Si Si Si Si 30 Si

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Chapter 9: Phase Diagrams
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